

**Supplementary material to: SCAN-based hybrid and double-hybrid density functionals
from models without fitted parameters**

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TABLES

TABLE I: Nonhydrogen transfer barrier heights (in kcal/mol)
of the NHTBH38/04 set [1].

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Heavy-atom transfer reactions											
H + N ₂ O → OH + N ₂											
V^f	18.14	9.97	9.57	13.91	12.08	17.41	15.34	20.83	19.30	23.58	22.66
V^r	83.22	52.46	64.71	68.79	76.54	76.63	81.26	79.93	82.69	81.25	83.28
H + FH → HF + H											
V^f	42.18	27.07	28.83	33.74	33.42	38.71	37.24	41.88	40.16	43.35	41.87
V^r	42.18	27.07	28.83	33.74	33.42	38.71	37.24	41.88	40.16	43.35	41.87
H + ClH → HCl + H											
V^f	18.00	9.64	9.51	13.38	11.59	16.03	13.85	18.06	16.14	19.26	17.78
V^r	18.00	9.64	9.51	13.38	11.59	16.03	13.85	18.06	16.14	19.26	17.78
H + FCH ₃ → HF + CH ₃											
V^f	30.38	18.51	20.38	25.71	25.25	30.23	28.79	32.79	31.06	33.87	32.35
V^r	57.02	40.99	46.44	49.75	53.03	54.86	56.60	57.43	58.24	58.38	58.85
H + F ₂ → HF + F											
V^f	2.27	-9.86	-11.36	-4.60	-8.46	-0.44	-4.25	5.79	2.74	11.20	8.89
V^r	106.18	80.86	89.93	99.03	103.53	109.15	111.27	116.64	117.43	121.25	121.59

Table I: NHTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
$\text{CH}_3 + \text{FCl} \longrightarrow \text{CH}_3\text{F} + \text{Cl}$											
V^f	7.43	-6.47	-4.62	0.79	1.39	4.97	5.08	7.86	7.84	10.02	10.03
V^r	60.17	41.63	45.99	53.46	55.24	60.14	60.54	64.14	63.88	66.55	66.21
Nucleophilic substitution reactions											
$\text{F}^- + \text{CH}_3\text{F} \longrightarrow \text{FCH}_3 + \text{F}^-$											
V^f	-0.34	-8.31	-8.37	-2.67	-3.20	-0.70	-1.16	-0.07	-0.40	-0.00	-0.17
V^r	-0.34	-8.31	-8.37	-2.67	-3.20	-0.70	-1.16	-0.07	-0.40	-0.00	-0.17
$\text{F}^- \cdots \text{CH}_3\text{F} \longrightarrow \text{FCH}_3 \cdots \text{F}^-$											
V^f	13.38	7.06	8.03	11.69	12.24	13.57	13.94	14.25	14.48	14.40	14.58
V^r	13.38	7.06	8.03	11.69	12.24	13.57	13.94	14.25	14.48	14.40	14.58
$\text{Cl}^- + \text{CH}_3\text{Cl} \longrightarrow \text{ClCH}_3 + \text{Cl}^-$											
V^f	3.10	-3.80	-5.10	0.87	-0.53	2.92	1.92	3.89	3.27	4.32	3.91
V^r	3.10	-3.80	-5.10	0.87	-0.53	2.92	1.92	3.89	3.27	4.32	3.91
$\text{Cl}^- \cdots \text{CH}_3\text{Cl} \longrightarrow \text{ClCH}_3 \cdots \text{Cl}^-$											
V^f	13.61	7.15	6.95	11.29	10.88	13.41	13.13	14.58	14.42	15.14	15.04
V^r	13.61	7.15	6.95	11.29	10.88	13.41	13.13	14.58	14.42	15.14	15.04
$\text{F}^- + \text{CH}_3\text{Cl} \longrightarrow \text{FCH}_3 + \text{Cl}^-$											
V^f	-12.54	-19.52	-21.98	-15.45	-17.67	-13.75	-15.21	-12.68	-13.56	-12.13	-12.64

Table I: NHTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
V^r	20.11	12.31	15.14	19.44	20.56	21.77	22.08	22.10	22.05	21.83	21.69
$F^- \cdots CH_3Cl \longrightarrow FCH_3 \cdots Cl^-$											
V^f	2.89	-0.67	-1.65	1.66	0.94	2.95	2.56	3.82	3.63	4.29	4.20
V^r	29.62	21.45	25.09	28.35	30.19	30.86	31.73	31.47	31.80	31.38	31.51
$OH^- + CH_3F \longrightarrow HOCH_3 + F^-$											
V^f	-2.78	-10.59	-10.24	-4.97	-5.24	-2.97	-3.27	-2.34	-2.56	-2.20	-2.34
V^r	17.33	9.14	9.15	15.94	15.34	18.22	17.64	18.70	18.23	18.50	18.22
$OH^- \cdots CH_3F \longrightarrow HOCH_3 \cdots F^-$											
V^f	10.96	3.89	5.16	8.82	9.54	10.87	11.35	11.68	11.99	11.94	12.16
V^r	47.2	43.21	44.02	49.06	49.29	51.09	51.18	51.44	51.47	51.22	51.22
Unimolecular and association reactions											
$H + N_2 \longrightarrow HN_2$											
V^f	14.69	5.19	4.35	8.50	6.40	11.54	9.42	14.77	13.19	17.32	16.25
V^r	10.72	9.08	9.70	11.64	11.75	12.55	12.48	12.41	12.32	11.91	11.71
$H + CO \longrightarrow HCO$											
V^f	3.17	-1.69	-3.48	0.31	-2.18	1.83	-0.41	3.08	1.37	3.85	2.59
V^r	22.68	24.66	24.12	25.47	24.68	25.51	24.71	25.03	24.42	24.52	24.10
$H + C_2H_4 \longrightarrow CH_3CH_2$											

Table I: NHTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
V^f	1.72	-0.14	-4.01	0.67	-3.68	1.50	-1.73	2.89	0.76	4.24	2.80
V^r	41.75	40.20	43.00	44.12	45.16	45.72	46.03	46.14	46.31	46.18	46.38
$\text{CH}_3 + \text{C}_2\text{H}_4 \longrightarrow \text{CH}_3\text{CH}_2\text{CH}_2$											
V^f	6.85	1.50	0.93	4.02	2.47	5.24	4.28	6.64	6.15	7.93	7.67
V^r	32.97	29.76	30.62	34.47	33.72	36.89	35.78	38.33	37.45	39.12	38.50
$\text{HCN} \longrightarrow \text{HCN}$											
V^f	48.16	45.60	46.27	46.37	46.47	47.30	47.11	48.45	48.18	49.35	49.09
V^r	33.11	30.50	32.14	32.51	33.42	33.31	33.69	33.71	33.79	33.92	33.90
MSE	-8.52	-7.48	-3.13	-3.28	-0.32	-0.83	1.39	0.87	2.34	1.94	
MAE	8.62	7.62	3.63	3.84	1.57	2.24	1.62	1.63	2.44	2.03	
rms	10.61	8.72	4.63	4.63	2.19	2.82	2.58	2.55	3.71	3.43	
Max(-)	-30.76	-18.51	-14.43	-10.73	-6.59	-6.52	-3.29	-2.02	-1.97	-0.58	
Max(+)	1.98	1.44	2.79	3.41	3.97	5.09	10.46	11.25	15.07	15.41	

TABLE II: Hydrogen transfer barrier heights (in kcal/mol) of the HTBH38/04 set [1, 2].

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
$\text{H} + \text{HCl} \longrightarrow \text{H}_2 + \text{Cl}$											
V^f	5.70	0.56	-1.05	2.94	0.31	4.88	2.18	6.56	4.26	7.64	5.83

Table II: HTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
V^r	8.70	-1.95	0.09	1.66	2.39	3.90	4.23	5.26	5.20	5.90	5.79
OH + H ₂ → H + H ₂ O											
V^f	5.10	-6.36	-2.01	-0.02	1.98	3.16	4.11	4.79	4.95	5.51	5.45
V^r	21.20	13.69	11.36	17.31	14.02	20.24	17.06	23.24	20.58	25.39	23.30
CH ₃ + H ₂ → H + CH ₄											
V^f	12.10	3.82	7.47	6.89	8.99	8.83	10.21	10.13	10.81	10.84	11.26
V^r	15.30	9.32	7.36	11.90	9.18	13.92	11.47	15.64	13.55	16.72	15.05
OH + CH ₄ → CH ₃ + H ₂ O											
V^f	6.70	-5.65	-1.58	1.62	3.48	4.91	5.87	6.17	6.44	6.50	6.53
V^r	19.60	8.91	11.90	13.94	15.33	16.90	17.56	19.10	19.34	20.50	20.59
H + H ₂ → H ₂ + H											
V^f	9.60	3.64	2.53	5.63	3.58	7.46	5.59	9.15	7.56	10.24	9.03
V^r	9.60	3.64	2.53	5.63	3.58	7.46	5.59	9.15	7.56	10.24	9.03
OH + NH ₃ → H ₂ O + NH ₂											

Table II: HTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
V^f	3.20	-11.94	-7.40	-2.28	-0.07	2.16	3.31	3.97	4.43	4.60	4.78
V^r	12.70	-0.71	3.31	7.41	9.37	11.49	12.47	13.73	14.13	14.91	15.07
HCl + CH ₃ → Cl + CH ₄											
V^f	1.70	-5.94	-3.08	-2.49	-0.90	-0.67	0.21	0.30	0.67	0.74	0.91
V^r	7.90	-2.95	-2.04	1.25	1.38	3.45	3.51	4.51	4.34	4.88	4.66
OH + C ₂ H ₆ → H ₂ O + C ₂ H ₅											
V^f	3.40	-9.03	-4.49	-1.23	0.77	2.22	3.37	3.58	3.96	4.00	4.08
V^r	19.90	10.75	13.13	15.65	16.62	18.30	18.71	20.25	20.34	21.52	21.51
F + H ₂ → HF + H											
V^f	1.80	-12.97	-7.77	-4.90	-2.56	-0.73	0.29	1.48	1.67	2.44	2.44
V^r	33.40	25.12	22.61	28.40	24.89	31.46	28.03	35.11	32.33	37.88	35.75
O + CH ₄ → OH + CH ₃											
V^f	13.70	-0.79	1.76	7.16	7.98	11.53	11.82	13.66	13.56	14.42	14.23
V^r	8.10	-0.57	3.31	3.94	6.33	6.48	7.86	8.07	8.83	8.93	9.43

Table II: HTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
$\text{H} + \text{PH}_3 \longrightarrow \text{PH}_2 + \text{H}_2$											
V^f	3.10	-1.78	-2.65	0.35	-1.67	1.86	-0.08	3.02	1.28	3.70	2.29
V^r	23.20	17.91	19.24	20.12	20.47	21.43	21.69	22.31	22.36	22.79	22.82
$\text{H} + \text{HO} \longrightarrow \text{H}_2 + \text{O}$											
V^f	10.70	3.74	3.39	7.13	5.69	9.66	7.81	11.87	10.16	13.34	11.94
V^r	13.10	-1.98	1.94	5.33	7.14	9.62	10.52	11.95	12.15	12.94	12.95
$\text{H} + \text{H}_2\text{S} \longrightarrow \text{H}_2 + \text{HS}$											
V^f	3.50	-1.22	-2.43	1.05	-1.10	2.69	0.56	3.95	2.10	4.69	3.21
V^r	17.30	9.10	11.25	12.06	13.02	13.75	14.34	14.70	14.88	15.12	15.20
$\text{O} + \text{HCl} \longrightarrow \text{OH} + \text{Cl}$											
V^f	9.80	-10.54	-3.88	2.22	4.86	8.05	9.21	11.10	11.44	12.38	12.46
V^r	10.40	-7.33	-1.30	2.74	5.48	7.12	8.55	9.71	10.39	11.04	11.42
$\text{NH}_2 + \text{CH}_3 \longrightarrow \text{CH}_4 + \text{NH}$											
V^f	8.00	0.71	4.52	5.05	7.38	7.26	8.59	8.44	9.15	9.00	9.44
V^r	22.40	10.56	12.29	16.42	16.77	19.62	19.63	21.32	21.03	22.03	21.69

Table II: HTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
$\text{NH}_2 + \text{C}_2\text{H}_5 \longrightarrow \text{C}_2\text{H}_6 + \text{NH}$											
V^f	7.50	2.89	5.99	6.88	8.82	8.71	9.85	9.53	10.16	9.86	10.27
V^r	18.30	7.50	9.61	13.69	14.21	16.98	17.24	18.67	18.55	19.38	19.15
$\text{C}_2\text{H}_6 + \text{NH}_2 \longrightarrow \text{NH}_3 + \text{C}_2\text{H}_5$											
V^f	10.40	1.43	5.26	7.25	9.01	9.81	10.88	10.69	11.07	10.87	10.97
V^r	17.40	9.98	12.17	14.45	15.41	16.57	17.06	17.60	17.76	18.08	18.12
$\text{NH}_2 + \text{CH}_4 \longrightarrow \text{CH}_3 + \text{NH}_3$											
V^f	14.50	4.39	7.85	9.87	11.51	12.37	13.25	13.29	13.55	13.49	13.53
V^r	17.80	7.71	10.62	12.50	13.92	15.03	15.78	16.47	16.75	17.18	17.30
$s\text{-trans } cis\text{-C}_5\text{H}_8 \longrightarrow s\text{-trans } cis\text{-C}_5\text{H}_8$											
V^f	38.40	31.20	33.63	35.58	37.16	37.35	38.19	37.23	37.54	36.46	36.57
V^r	38.40	31.20	33.63	35.58	37.16	37.35	38.19	37.23	37.54	36.46	36.57

Table II: HTBH38/04 (continued).

Reaction	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
	MSE	-9.67	-7.49	-4.60	-3.99	-1.87	-1.81	-0.28	-0.56	0.50	0.19
	MSE	9.67	7.49	4.60	4.06	1.93	2.00	0.99	1.14	1.39	1.18
	rms	10.37	7.94	4.88	4.46	2.19	2.48	1.30	1.43	1.74	1.45
	Max(-)	-20.34	-13.68	-7.77	-8.51	-4.80	-5.37	-3.44	-3.56	-3.02	-3.24
	Max(+)	-4.61	-1.51	-0.62	1.32	1.21	2.35	2.04	2.66	4.48	2.77

TABLE III: Interaction energies (in kcal/mol) of the S22 set [3, 4].

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Hydrogen bonded complexes											
NH ₃ dimer (C2h)	-3.17	-2.83	-3.14	-2.77	-3.11	-2.83	-3.15	-2.90	-3.15	-2.93	-3.12
H ₂ O dimer (Cs)	-5.02	-4.91	-5.39	-4.92	-5.38	-4.96	-5.34	-4.94	-5.22	-4.89	-5.09
Formic acid dimer (C2h)	-18.80	-18.08	-20.63	-18.86	-20.75	-19.21	-20.49	-19.08	-19.87	-18.75	-19.28
Formamide dimer (C2h)	-16.12	-14.69	-16.38	-15.27	-16.59	-15.64	-16.60	-15.69	-16.34	-15.55	-16.01
Uracil dimer (C2h)	-20.69	-18.45	-20.32	-19.16	-20.65	-19.72	-20.81	-19.95	-20.68	-19.91	-20.43
2-pyridone ... 2-aminopyridine (C1)	-17.00	-15.27	-16.69	-15.35	-16.53	-15.84	-16.73	-16.26	-16.88	-16.42	-16.86
Adenine ... thymine WC (C1)	-16.74	-14.29	-15.88	-14.50	-15.83	-15.09	-16.10	-15.54	-16.25	-15.69	-16.17
Dispersion complexes											

Table III: S22 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
CH ₄ dimer (D3d)	-0.53	-0.08	-0.37	-0.04	-0.31	-0.09	-0.32	-0.18	-0.36	-0.25	-0.38
C ₂ H ₄ dimer (D2d)	-1.50	-0.31	-1.07	-0.35	-1.04	-0.58	-1.15	-0.84	-1.26	-1.00	-1.31
Benzene ... CH ₄	-1.45	-0.01	-0.89	-0.08	-0.82	-0.41	-0.97	-0.79	-1.19	-1.04	-1.33
Benzene dimer (C2h)	-2.62	1.90	-1.13	1.80	-0.57	0.60	-1.03	-0.91	-1.95	-1.94	-2.66
Pyrazine dimer (Cs)	-4.20	0.73	-2.71	0.53	-2.14	-0.82	-2.64	-2.45	-3.61	-3.57	-4.35
Uracil dimer (C2)	-9.74	-2.69	-8.00	-3.30	-7.32	-4.99	-7.76	-6.81	-8.57	-7.91	-9.14
Indole Benzene (C1)	-4.59	2.24	-2.19	2.02	-1.39	0.17	-2.17	-2.12	-3.59	-3.67	-4.68
Adenine ... thymine (C1)	-11.66	-1.32	-8.69	-2.14	-7.73	-4.86	-8.65	-7.83	-10.21	-9.69	-11.32
Mixed complexes											
Ethene ... Ethine (C2v)	-1.51	-1.16	-1.35	-1.18	-1.38	-1.26	-1.46	-1.36	-1.51	-1.41	-1.53
Benzene ... H ₂ O (Cs)	-3.29	-2.04	-3.30	-2.21	-3.19	-2.54	-3.23	-2.83	-3.28	-2.98	-3.30
Benzene ... NH ₃ (Cs)	-2.32	-0.92	-2.00	-1.03	-1.89	-1.36	-1.99	-1.72	-2.14	-1.94	-2.24
Benzene ... HCN (Cs)	-4.55	-2.81	-4.08	-3.28	-4.29	-3.83	-4.55	-4.25	-4.73	-4.46	-4.79
Benzene dimer (C2v)	-2.71	-0.10	-1.50	-0.28	-1.43	-0.95	-1.79	-1.70	-2.28	-2.19	-2.60
Indole ... Benzene (Cs)	-5.62	-2.03	-4.07	-2.36	-3.97	-3.32	-4.48	-4.37	-5.14	-5.03	-5.57
Phenol dimer (C1)	-7.09	-3.87	-5.92	-4.18	-5.83	-4.96	-6.18	-5.76	-6.58	-6.23	-6.82

Table III: S22 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
	MSE	2.72	0.69	2.45	0.85	1.75	0.60	1.03	0.28	0.61	0.09
	MAE	2.72	0.92	2.46	1.11	1.78	0.84	1.05	0.43	0.61	0.19
	rms	3.73	1.22	3.45	1.54	2.49	1.19	1.42	0.59	0.78	0.25
	Max(-)	0.11	-1.83	-0.06	-1.95	-0.41	-1.69	-0.28	-1.07	0.05	-0.48
	Max(+)	10.34	2.97	9.52	3.93	6.80	3.01	3.83	1.45	1.97	0.60

TABLE IV: Interaction energies (in kcal/mol) of the S66 set [5].

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Hydrogen bonded complexes											
Water ... Water	-4.918	-4.87	-5.32	-4.90	-5.33	-4.94	-5.30	-4.93	-5.20	-4.88	-5.08
Water ... MeOH	-5.592	-5.16	-5.95	-5.20	-5.88	-5.33	-5.86	-5.42	-5.79	-5.43	-5.70
Water ... MeNH2	-6.908	-7.15	-7.46	-7.01	-7.36	-7.01	-7.33	-6.99	-7.24	-6.94	-7.13
Water ... Peptide	-8.103	-7.28	-8.68	-7.49	-8.63	-7.72	-8.56	-7.83	-8.39	-7.82	-8.21
MeOH ... MeOH	-5.757	-5.08	-5.89	-5.10	-5.80	-5.26	-5.82	-5.40	-5.80	-5.46	-5.75
MeOH ... MeNH2	-7.554	-7.19	-7.83	-7.05	-7.67	-7.16	-7.69	-7.28	-7.68	-7.32	-7.62
MeOH ... Peptide	-8.230	-7.18	-8.53	-7.30	-8.43	-7.56	-8.42	-7.74	-8.33	-7.78	-8.20
MeOH ... Water	-5.009	-4.78	-5.27	-4.79	-5.26	-4.85	-5.26	-4.88	-5.18	-4.87	-5.09
MeNH2 ... MeOH	-3.059	-2.22	-2.76	-2.17	-2.68	-2.31	-2.76	-2.50	-2.84	-2.61	-2.86

Table IV: S66 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
MeNH2 ... MeNH2	-4.160	-3.08	-3.87	-3.01	-3.75	-3.24	-3.86	-3.51	-3.97	-3.67	-4.01
MeNH2 ... Peptide	-5.419	-3.39	-4.96	-3.53	-4.84	-3.98	-4.98	-4.43	-5.12	-4.67	-5.17
MeNH2 ... Water	-7.266	-7.14	-7.87	-7.06	-7.72	-7.15	-7.68	-7.22	-7.59	-7.22	-7.49
Peptide ... MeOH	-6.187	-4.53	-5.69	-4.72	-5.69	-5.09	-5.85	-5.44	-5.98	-5.63	-6.02
Peptide ... MeNH2	-7.454	-6.12	-7.06	-6.12	-6.96	-6.44	-7.12	-6.78	-7.27	-6.96	-7.32
Peptide ... Peptide	-8.630	-6.36	-8.00	-6.64	-7.98	-7.16	-8.16	-7.62	-8.31	-7.84	-8.34
Peptide ... Water	-5.124	-4.32	-4.88	-4.46	-4.98	-4.64	-5.08	-4.78	-5.11	-4.83	-5.08
Uracil ... Uracil (BP)	-17.182	-15.49	-17.25	-15.98	-17.41	-16.47	-17.53	-16.70	-17.42	-16.69	-17.20
Water ... Pyridine	-6.857	-6.82	-7.13	-6.64	-7.01	-6.69	-7.02	-6.75	-7.01	-6.76	-6.96
MeOH ... Pyridine	-7.410	-6.92	-7.42	-6.73	-7.24	-6.86	-7.31	-7.04	-7.38	-7.14	-7.39
AcOH ... AcOH	-19.093	-18.59	-21.09	-19.24	-21.12	-19.58	-20.87	-19.49	-20.29	-19.20	-19.73
AcNH2 ... AcNH2	-16.265	-15.02	-16.64	-15.48	-16.78	-15.85	-16.81	-15.94	-16.59	-15.85	-16.30
AcOH ... Uracil	-19.491	-18.31	-20.45	-18.96	-20.62	-19.39	-20.56	-19.45	-20.21	-19.28	-19.81
AcNH2 ... Uracil	-19.189	-17.67	-19.53	-18.31	-19.78	-18.79	-19.85	-18.93	-19.64	-18.83	-19.33
Dispersion complexes											
Benzene ... Benzene (pi-pi)	-2.822	1.22	-1.34	1.18	-0.83	0.18	-1.24	-1.13	-2.05	-2.04	-2.68
Pyridine ... Pyridine (pi-pi)	-3.895	0.48	-2.39	0.34	-1.90	-0.79	-2.34	-2.20	-3.20	-3.17	-3.86
Uracil ... Uracil (pi-pi)	-9.829	-2.51	-8.00	-3.15	-7.30	-4.91	-7.75	-6.77	-8.58	-7.90	-9.15

Table IV: S66 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Benzene ... Pyridine (pi-pi)	-3.439	0.81	-1.94	0.71	-1.44	-0.37	-1.87	-1.74	-2.71	-2.68	-3.35
Benzene ... Uracil (pi-pi)	-5.713	0.36	-3.81	0.01	-3.18	-1.49	-3.69	-3.22	-4.62	-4.34	-5.30
Pyridine ... Uracil (pi-pi)	-6.819	-0.74	-4.97	-1.12	-4.37	-2.62	-4.86	-4.34	-5.76	-5.44	-6.42
Benzene ... Ethene	-1.432	0.76	-0.81	0.82	-0.45	0.32	-0.60	-0.36	-0.97	-0.84	-1.27
Uracil ... Ethene	-3.380	-0.34	-2.56	-0.52	-2.27	-1.22	-2.47	-2.01	-2.83	-2.51	-3.08
Uracil ... Ethyne	-3.738	-1.03	-3.16	-1.24	-2.92	-1.88	-3.06	-2.57	-3.34	-3.00	-3.54
Pyridine ... Ethene	-1.872	0.45	-1.24	0.48	-0.88	-0.06	-1.04	-0.77	-1.43	-1.27	-1.73
Pentane ... Pentane	-3.776	0.42	-1.90	0.38	-1.61	-0.45	-2.02	-1.46	-2.59	-2.12	-2.95
Neopentane ... Pentane	-2.613	0.18	-0.98	0.23	-0.82	-0.26	-1.14	-0.89	-1.55	-1.31	-1.80
Neopentane ... Neopentane	-1.777	0.14	-1.16	0.18	-0.99	-0.35	-1.31	-1.01	-1.72	-1.45	-1.98
Cyclopentane ... Neopentane	-2.404	0.11	-1.25	0.18	-1.07	-0.41	-1.45	-1.18	-1.96	-1.69	-2.27
Cyclopentane ... Cyclopentane	-2.997	0.05	-0.59	0.16	-0.48	-0.16	-0.73	-0.59	-1.04	-0.89	-1.24
Benzene ... Cyclopentane	-3.575	0.30	-1.96	0.18	-1.66	-0.72	-2.07	-1.80	-2.72	-2.52	-3.17
Benzene ... Neopentane	-2.895	0.09	-1.57	0.00	-1.36	-0.67	-1.70	-1.49	-2.20	-2.04	-2.54
Uracil ... Pentane	-4.848	0.52	-2.89	0.27	-2.43	-0.91	-2.88	-2.25	-3.58	-3.09	-4.04
Uracil ... Cyclopentane	-4.138	0.58	-2.25	0.44	-1.83	-0.58	-2.24	-1.77	-2.90	-2.54	-3.35
Uracil ... Neopentane	-3.712	0.08	-2.26	-0.07	-1.93	-0.87	-2.24	-1.80	-2.73	-2.39	-3.06
Ethene ... Pentane	-2.005	0.01	-1.12	0.03	-0.98	-0.36	-1.18	-0.85	-1.45	-1.17	-1.62
Ethyne ... Pentane	-1.748	-0.09	-1.14	-0.06	-0.96	-0.39	-1.10	-0.82	-1.33	-1.12	-1.49

Table IV: S66 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Peptide ... Pentane	-4.264	0.07	-2.69	-0.07	-2.34	-0.97	-2.68	-2.00	-3.19	-2.65	-3.51
Mixed complexes											
Benzene ... Benzene (TS)	-2.876	-0.19	-1.66	-0.36	-1.56	-1.03	-1.92	-1.79	-2.40	-2.29	-2.72
Pyridine ... Pyridine (TS)	-3.535	-0.75	-2.31	-0.97	-2.26	-1.68	-2.64	-2.46	-3.11	-2.95	-3.42
Benzene ... Pyridine (TS)	-3.331	-0.61	-2.13	-0.84	-2.08	-1.53	-2.45	-2.30	-2.91	-2.78	-3.22
Benzene ... Ethyne (CH-pi)	-2.867	-1.23	-2.31	-1.48	-2.36	-1.94	-2.59	-2.38	-2.82	-2.64	-2.95
Ethyne ... Ethyne (TS)	-1.524	-1.20	-1.38	-1.21	-1.41	-1.28	-1.47	-1.35	-1.51	-1.39	-1.51
Benzene ... AcOH (OH-pi)	-4.707	-2.32	-4.31	-2.65	-4.16	-3.28	-4.31	-3.86	-4.52	-4.18	-4.63
Benzene ... AcNH2 (NH-pi)	-4.361	-2.34	-3.82	-2.59	-3.78	-3.10	-3.98	-3.59	-4.18	-3.86	-4.28
Benzene ... Water (OH-pi)	-3.277	-1.98	-3.33	-2.17	-3.21	-2.52	-3.25	-2.82	-3.30	-2.98	-3.31
Benzene ... MeOH (OH-pi)	-4.188	-1.86	-3.75	-2.03	-3.52	-2.61	-3.66	-3.21	-3.91	-3.58	-4.06
Benzene ... MeNH2 (NH-pi)	-3.231	-0.77	-2.52	-0.90	-2.30	-1.48	-2.50	-2.13	-2.81	-2.55	-3.03
Benzene ... Peptide (NH-pi)	-5.282	-1.68	-4.05	-2.01	-3.87	-2.90	-4.23	-3.86	-4.73	-4.45	-5.06
Pyridine ... Pyridine (CH-N)	-4.146	-2.49	-3.01	-2.60	-3.15	-3.02	-3.51	-3.43	-3.81	-3.66	-3.95
Ethyne ... Water (CH-O)	-2.850	-2.63	-2.82	-2.74	-2.96	-2.81	-3.03	-2.83	-3.01	-2.81	-2.95
Ethyne ... AcOH (OH-pi)	-4.868	-4.10	-5.23	-4.30	-5.21	-4.50	-5.18	-4.61	-5.07	-4.62	-4.94
Pentane ... AcOH	-2.912	0.07	-1.82	0.02	-1.54	-0.57	-1.76	-1.26	-2.10	-1.71	-2.31
Pentane ... AcNH2	-3.534	-0.26	-2.35	-0.32	-2.05	-0.98	-2.31	-1.75	-2.68	-2.23	-2.91

Table IV: S66 interaction energies (continued).

Complex	Reference	PBE	SCAN	PBE0	SCAN0	PBE0-DH	SCAN0-DH	PBE-QIDH	SCAN-QIDH	PBE0-2	SCAN0-2
Benzene ... AcOH	-3.801	-0.53	-2.83	-0.81	-2.61	-1.60	-2.88	-2.46	-3.29	-2.99	-3.57
Peptide ... Ethene	-2.999	-0.95	-2.34	-1.06	-2.22	-1.49	-2.38	-1.96	-2.58	-2.25	-2.70
Pyridine ... Ethyne	-3.991	-3.64	-3.73	-3.65	-3.82	-3.78	-3.97	-3.89	-4.06	-3.94	-4.07
MeNH2 ... Pyridine	-3.968	-1.73	-3.30	-1.73	-3.05	-2.25	-3.24	-2.88	-3.56	-3.28	-3.77
	MSE	2.22	0.61	2.09	0.77	1.56	0.58	1.00	0.33	0.66	0.19
	MAE	2.23	0.85	2.10	1.01	1.58	0.81	1.01	0.49	0.67	0.27
	rms	2.75	1.04	2.61	1.25	1.98	1.01	1.27	0.63	0.83	0.40
	Max(-)	-0.24	-2.00	-0.15	-2.03	-0.49	-1.77	-0.40	-1.20	-0.10	-0.64
	Max(+)	7.32	2.41	6.68	2.53	4.92	2.27	3.06	1.95	2.11	1.76